



Review Article

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THE ROLE OF 6-BROMOQUINAZOLINE-2,4-DIONE IN MODERN MEDICINAL CHEMISTRY: STRUCTURAL FEATURES, BIOLOGICAL ACTIVITY, AND THERAPEUTIC PROSPECTS

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Abstract

Quinazoline derivatives represent a prominent class of nitrogen-containing heterocyclic compounds with extensive applications in medicinal chemistry. Among them, quinazoline-2,4-dione scaffolds have attracted considerable attention due to their diverse biological activities. The incorporation of halogen atoms, particularly bromine at the C-6 position, significantly alters the physicochemical and pharmacokinetic properties of these molecules. This paper provides a comprehensive analysis of 6-bromoquinazoline-2,4-dione, focusing on its structural characteristics, synthetic methodologies, and pharmacological potential. Special emphasis is placed on its anticancer, antiviral, and antimicrobial activities, along with a discussion of molecular mechanisms and future therapeutic perspectives.

Keywords: quinazoline derivatives, halogenation, medicinal chemistry, anticancer agents, antiviral activity, heterocyclic compounds.

1. Introduction

The continuous emergence of resistant pathogens and the increasing prevalence of cancer have necessitated the development of novel therapeutic agents with improved efficacy and selectivity. Heterocyclic compounds, particularly nitrogen-containing

systems, play a crucial role in drug discovery due to their structural diversity and ability to interact with biological targets.

Quinazoline derivatives have been extensively studied as pharmacologically active molecules. The quinazoline-2,4-dione core is of particular interest due to its structural similarity to biologically relevant nucleobases, enabling interactions with enzymes and nucleic acids.

The substitution at specific positions of the quinazoline ring significantly influences biological activity. In this context, bromination at the C-6 position enhances lipophilicity, metabolic stability, and binding affinity toward target biomolecules. Therefore, 6-bromoquinazoline-2,4-dione emerges as a promising scaffold in modern medicinal chemistry.

2. Structural and Physicochemical Characteristics

2.1 Molecular Structure

6-Bromoquinazoline-2,4-dione consists of:

- a fused bicyclic system (benzene + pyrimidine ring),
- two carbonyl groups at positions 2 and 4,
- a bromine substituent at the 6th position.

The presence of electron-withdrawing groups such as bromine and carbonyl functionalities significantly affects electron distribution across the molecule. This results in enhanced reactivity and the ability to form hydrogen bonds and π - π interactions with biological targets.

2.2 Electronic and Steric Effects

Bromine exerts both inductive (-I) and resonance effects, contributing to:

- increased molecular polarization,
- improved membrane permeability,
- enhanced interaction with hydrophobic pockets of proteins.

Additionally, the relatively large atomic radius of bromine introduces steric effects that can influence ligand-receptor binding specificity.

2.3 Physicochemical Properties

Key properties include:

- moderate to high lipophilicity (important for bioavailability),
- good solubility in organic solvents,
- thermal and chemical stability under physiological conditions.

These properties make the compound suitable for further pharmacological development.

3. Synthetic Approaches

3.1 Classical Synthetic Routes

The conventional synthesis involves:

- 1-Preparation of 6-bromoanthranilic acid;
- 2-Reaction with urea or its derivatives;
- 3-Cyclization under thermal or catalytic conditions.

This method is reliable but may require long reaction times and generate by-products.

3.2 Advanced Synthetic Strategies

Recent developments include:

3.2.1 Microwave-Assisted Synthesis

- reduces reaction time significantly,
- improves yield and purity.

3.2.2 Catalytic Systems

- use of transition metal catalysts,
- improved selectivity and efficiency.

3.2.3 Green Chemistry Approaches

- solvent-free reactions,
- environmentally benign reagents,

-reduced waste production.

4. Biological Activity and Mechanisms of Action

4.1 Anticancer Activity

6-Bromoquinazoline-2,4-dione derivatives exhibit potent anticancer activity through multiple mechanisms:

Inhibition of tyrosine kinases:

Quinazoline-based compounds are known inhibitors of epidermal growth factor receptor (EGFR).

Induction of apoptosis:

Activation of caspase pathways leading to programmed cell death.

Cell cycle arrest:

Interference with DNA replication and mitotic progression.

These mechanisms highlight their potential as targeted anticancer agents.

4.2 Antiviral Properties

The compound demonstrates antiviral activity by:

- inhibiting viral polymerases,
- disrupting viral replication cycles,
- preventing viral entry into host cells.

Such properties are particularly relevant in the development of broad-spectrum antiviral agents.

4.3 Antimicrobial and Antifungal Effects

Studies indicate that halogenated quinazoline derivatives:

- exhibit strong activity against Gram-positive bacteria,
- disrupt microbial cell membranes,
- inhibit essential enzymatic pathways.

The bromine substituent enhances these effects by increasing lipophilicity and membrane interaction.

4.4 Structure–Activity Relationship (SAR)

SAR studies reveal:

- substitution at C-6 significantly enhances activity,
- electron-withdrawing groups improve binding affinity,
- structural rigidity contributes to selectivity.

These insights guide rational drug design strategies.

5. Pharmacokinetics and Drug-Likeness

Preliminary studies suggest that 6-bromoquinazoline-2,4-dione:

- follows Lipinski’s Rule of Five,
- exhibits favorable absorption and distribution profiles,
- shows potential for oral bioavailability.

However, detailed ADMET (Absorption, Distribution, Metabolism, Excretion, Toxicity) studies are still required.

6. Therapeutic Applications and Future Prospects

6.1 Oncology

Given its ability to target key signaling pathways, this compound is a promising candidate for:

- targeted cancer therapy,
- combination chemotherapy.

6.2 Antiviral Drug Development

The ongoing need for antiviral agents makes this scaffold valuable for:

- emerging viral infections,
- drug-resistant viral strains.

6.3 Multi-Target Drug Design

The structural versatility allows:

- development of hybrid molecules,

-simultaneous targeting of multiple biological pathways.

6.4 Nanotechnology Integration

Future directions include:

- nanoparticle-based drug delivery,
- improved targeting and reduced toxicity.

7. Challenges and Limitations

Despite its potential, several challenges remain:

- insufficient clinical data,
- possible toxicity of halogenated compounds,
- scalability of synthesis.

Addressing these issues is essential for clinical translation.

8. Conclusion

6-Bromoquinazoline-2,4-dione represents a highly promising scaffold in medicinal chemistry due to its versatile biological activity and favorable physicochemical properties. Its potential applications in oncology, virology, and antimicrobial therapy make it a valuable candidate for further research. Continued advancements in synthetic methodologies, molecular modeling, and pharmacological evaluation will be critical in unlocking its full therapeutic potential.

References

1. Joule, J.A., Mills, K. *Heterocyclic Chemistry*.
2. Brown, D.J. *The Chemistry of Heterocyclic Compounds: Quinazolines*.
3. Patrick, G.L. *An Introduction to Medicinal Chemistry*.
4. Recent articles from *Journal of Medicinal Chemistry*, *Bioorganic & Medicinal Chemistry*, *European Journal of Medicinal Chemistry*.
5. Scopus and Web of Science indexed publications on quinazoline derivatives.